

40 years of N₂O isotopomer research: Advances in analytics and exemplary applications

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Variations of nitrogen isotope ratios in N₂O in nature were first reported by Yoshida et al. 40 years ago [1,2]. Subsequent studies have developed analytical techniques for oxygen isotope ratios, intramolecular nitrogen isotope distribution (isotopomers), and molecules doubly substituted with heavy isotopes (clumped isotopes).

In the pioneering works, N₂O had to be converted to N₂ and O₂/CO₂ for elemental isotope ratio measurements on an isotope ratio mass spectrometer (IRMS) with double collectors. But modern IRMS equipped with multi-collector system and GC has enabled direct mass analysis of N₂O and its fragment NO from gas/water samples [3]. More recently, high-resolution IRMS allows analyses of clumped N₂O molecules and ¹⁷O/¹⁶O ratio that is a key parameter to assess non-mass dependent fractionation of three O isotopes [4].

Mid-IR laser spectroscopy has been established as a complementary approach, offering high temporal resolved data and selectivity for isotopomers. Depending on the target isotopic species and research question, analyzers employing different wavelength regions and detection schemes are at hand [5]. Using different analytical settings, applications ranging from on-line analysis in ambient air to high-precision analysis of clumped isotopic species have been realized [6].

Analysis of N₂O isotopomers/isotopologues/isotopocules has been applied to elucidate the global cycle and production/decomposition mechanisms of this greenhouse gas and ozone-depleting gas in the atmosphere, hydrosphere, and biosphere. We would like to review N₂O isotope studies to date and discuss current achievements and future challenges.

References

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